## STEREOSELECTIVE TOTAL SYNTHESES OF (±)-ZONAROL AND (±)-ISOZONAROL

## S. C. Welch and A. S. C. P. Rao Department of Chemistry University of Houston Houston, Texas 77004

(Received in USA 22 July 1976; received in UK for publication 10 January 1977) The two naturally occurring hydroquinones zonarol (1b) and isozonarol (2b) were isolated from brown seaweed <u>Dictyopteris</u> undulata found in the Pacific Ocean.<sup>1,2</sup> The structure and absolute stereochemistry of zonarol (1b) and isozonarol (2b) were rigorously defined by degradation and spectroscopy.<sup>1,2,3</sup> These marine natural products were found to be active against the following pathogenic fungi: Phytophthora cinnamoni, Rhizoctonia solani, Sclerotinia sclerotiorum, and sclerotium rolfsii.<sup>1</sup> We wish to report the first stereoselective and total syntheses of both  $(\pm)$ -zonarol  $(\underline{1b})$  and  $(\pm)$  isozonarol  $(\underline{2b})$ .



The starting material chosen for these syntheses is readily available ketol 3.4 Wolff-Kishner reduction of ketol 3 using hydrazine-hydrate in diethylene glycol (DEG) in the presence

of potassium hydroxide at 200° gives alcohol 4 in 90% yield. 5.6 Oxidation of alcohol 4 with Jones' reagent affords ketone 5 in 82% yield.<sup>7</sup> Addition of methyllithium in ether at  $0^{\circ}$  to ketone 5 produces tertiary alcohol 6 in 90% yield. When compound 6 was treated with standard reagents for dehydrating tertiary alcohols (SOC1<sub>2</sub>, pyr.; POC1<sub>2</sub>, pyr.; I<sub>2</sub>, benzene,  $\Delta$ ; or H<sub>2</sub>SO<sub>4</sub>, pentane) only substantial quantities of rearranged products were obtained. However, upon heating in anhydrous dimethyl sulfoxide at 155° for 16 hours, compound 6 is smoothly converted to alkene 7 in 77% yield.<sup>8</sup> Epoxidation of alkene 7 using meta-chloroperbenzoic acid in chloroform in the presence of disodium hydrogen phosphate gives a mixture of epoxides 8 in 98% yield. Epoxide ring opening by treatment of compounds 8 with lithium n-propylamide in tetrahydrofuran at reflux for 6 hours produces a mixture of allylic alcohols.<sup>9</sup> Oxidation of this mixture of allylic alcohols with Collins' reagent<sup>10</sup> afords enones 9a and  $9b^{11}$  (ratio 32 : 68, respectively) in 69% yield. Enones 9a and 9b are easily separated by column chromatography on E.Merck silica gel-60 using 15% ether: 85% pet-ether (bp 30-60°) as the eluant. Conjugate addition of 2,5-dimethoxyphenylmagnesium bromide Grignard reagent 12 in 1,2-dimethoxyethane (DME) with enone <u>9b</u> followed by quenching with freshly distilled acetic anhydride gives a crude enolacetate. Treatment of this crude enolacetate with potassium hydroxide in methanol produces ketone 10 (mp 108-109°) in 70% overall yield from enone 9b.<sup>13</sup>

Wittig reaction of ketone <u>10</u> with methylenetriphenylphosphorane in anhydrous dimethylsulfoxide at 80° for 24 hours affords (±)-zonarol dimethyl ether <u>(1a</u>, mp 117-118°) in 93% yield.<sup>14</sup> Treatment of ketone <u>10</u> with methyllithium in ether at 0° followed by dehydration of the resulting tertiary alcohol by heating in anhydrous dimethylsulfoxide at 155° for 16 hours gives (±)-zonarol dimethyl ether <u>(1a)</u> and (±)-isozonarol dimethyl ether <u>(2a)</u> (ratio 1 : 4.8, respectively) in 67% yield. Both zonarol dimethyl ether <u>(1a)</u> and isozonarol dimethyl ether <u>(2a)</u> were identical (ir, nmr, glc, and tlc) to the respective dimethyl ethers prepared from natural zonarol <u>(1b)</u> and isozonarol <u>(2b)</u>.<sup>15</sup> Synthetic ethers <u>1a</u> and <u>2a</u> were smoothly cleaved to (±)-zonarol <u>(1b)</u> and (±)-isozonarol <u>(2b)</u>, in 90 and 86% yield, respectively, by treatment with lithium n-butyl mercaptide in hexamethylphosphoramide at 150° for 24 hours.<sup>16</sup>

Acknowledgement: We graciously thank The Robert A. Welch Foundation (Grant No. E-518) and the Link Foundation for support of this research program.



a)  $N_2H_4$ , KOH, DEG,  $\Delta$ ; b)  $CrO_3$ ,  $H_2SO_4$ ,  $H_2O$ , acetone; c)  $CH_3Li$ ,  $Et_2O$ ,  $O^\circ$ ; d) DMSO, 155°; e) MCPBA,  $Na_2HPO_4$ ,  $CHC1_3$ ; f)  $LiN(nPr)_2$ , THF,  $\Delta$ , 6hr.; g)  $CrO_3 Pyr_2$ ,  $CH_2Cl_2$ ; h) chromatography, E.Merck, silica gel-60, 15%  $Et_2O$ : 85% pet-ether (bp 30-60°) eluant; i) 2,5-dimethoxyphenylmagnesium bromide, DME; j)  $Ac_2O$ ; k) KOH,  $CH_3OH$ .

## References

- 1. W. Fenical and O. McConnell, Experientia, <u>31</u>, 1004 (1975); W. Fenical, J. J. Sims, D. Squatrito, R. M. Wing, and P. Radlick, J. Org. Chem., <u>38</u>, 2383 (1973).
- 2. G. Cimino, P. DeLuca, S. DeStefano, and L. Minale, Tetrahedron, 31, 271 (1975).
- 3. Personal communication with Professor W. Fenical.
- 4. J. S. Dutcher, J. G. Macmillan, and C. H. Heathcock Tetrahedron Lett., 929 (1974).
- 5. D. Todd, Org. Reactions, <u>4</u>, 378 (1948).
- 6. All new compounds reported herein gave satisfactory elemental analyses.
- K. Bowden, I. M. Heilbron, E. R. H. Jones, and B. C. L. Weedon, J. Chem. Soc., 39 (1946);
  C. Djerassi, R. R. Engle, and A. Bowers, J. Org. Chem., <u>21</u>, 1547 (1956).
- V. J. T. Raynelis, W. L. Hergenrother, H. T. Hanson, and J. A. Valicenti, J. Org. Chem., 29, 123 (1969).
- 9. J. K. Crandall and L. C. Crawley, Org. Syntheses, 53, 17 (1973).
- R. Ratcliffe and R. Rodehorst, J. Org. Chem., <u>35</u>, 4000 (1970); J. C. Collins and W. W. Hess, Org. Syntheses, <u>52</u>, 5 (1972).
- Previously prepared via different synthetic routes by: E. Romann, A. J. Frey, P. A. Stadler, and A. Eschenmoser, Helv. Chim. Acta, <u>40</u>, 1900 (1957); R. W. Skeean, G. L. Trammell, and J. D. White, Tetrahedron Lett., 525 (1976).
- 12. D. F. MacSweeney and R. Ramage, Tetrahedron, 27, 1481 (1971).
- 13. R. E. Ireland, S. W. Baldwin, and S. C. Welch, J. Amer. Chem. Soc., 94, 2056 (1972).
- 14. A. Maercker, Org. Reactions, 14, 270 (1965).
- 15. We are grateful to Professor W. Fenical for providing natural samples of zonarone and isozonarone which were converted to the respective methyl ethers by reduction (NaBH<sub>4</sub>,  $CH_3OH$ ) and alkylation (CH<sub>3</sub>I, K<sub>2</sub>CO<sub>3</sub>, acetone).
- G. I. Feutrill and R. N. Mirrington, Aust. J. Chem., <u>25</u>, 1917, 1731 (1972); Tetrahedron Lett., 1327 (1970).